

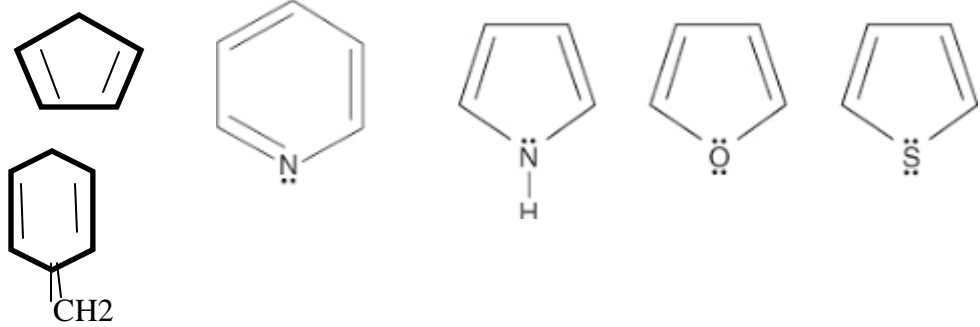


# INDIAN SCHOOL DARSAIT



## DEPARTMENT OF CHEMISTRY

|                            |   |  |  |                              |  |
|----------------------------|---|--|--|------------------------------|--|
| Subject: Chemistry         |   | Topic : ORGANIC CHEMISTRY AND HYDROCARBONS |  | Date of Worksheet: 5.12.2018 |  |
| Resource Person: Rohitha   |   | Date of Submission: _____                  |  |                              |  |
| Name of the Student: _____ |   | Class & Division: XI                       |  | Roll Number: _____           |  |
| 1                          | Write the IUPAC name of the following<br>a) $\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)_2$<br>b) $\text{CH}_3(\text{CH}_2)_4\underset{\text{CH}_2\text{CH}(\text{CH}_3)_2}{\text{CH}}\text{CH}_2\text{CH}_3$<br>c) $\text{CH}_3\text{CH}=\text{CHCH}_2\text{CH}_2\text{C}\equiv\text{CH}$<br>d) $\text{CH}_3\text{CH}_2\text{CH}(\text{Cl})\text{CH}_2\text{CH}(\text{OH})\text{CH}_2\text{COOH}$ |  |  | 1 each                       |  |
| 2                          | Write the structure of the following<br>a) 2-amino-3-hydroxypropanal<br>b) 3-phenylprop-2-en-1-oic acid<br>c) 2-methoxy-2,3-dimethylbutanal<br>d) 5,6-Diethyl-3-methyldec-4-ene<br>e) 2-Ethyl-3-hydroxy-2-methylpropanal  |  |  | 1 each                       |  |
| 3                          | Write all the possible isomers of hexane with their names.  |  |  | 2                            |  |
| 4                          | Arrange the following in the increasing order of stability and justify your answer.<br>a) $\text{CH}_3\text{CH}_2^+$ , $(\text{CH}_3)_2\text{CH}^+$ , $\text{CH}_3^+$ , $(\text{CH}_3)_3\text{C}^+$<br>b) $\text{CH}_3^\cdot$ , $(\text{CH}_3)_3\text{C}^\cdot$ , $(\text{CH}_3)_2\text{CH}^\cdot$ , $\text{CH}_3\text{CH}_2^\cdot$   |  |  | 2 each                       |  |
| 5                          | a) Draw the resonance structures of $\text{C}_6\text{H}_5\text{OH}$ , $\text{C}_6\text{H}_5\text{CHO}$<br>b) Draw the hyper conjugative structures of 2-butene  |  |  | 2 each                       |  |
| 6                          | Give a short note on the acidic character of alkyne.  |  |  | 2                            |  |
| 7                          | Explain in brief<br>a) Dumas method<br>b) Carius method for Sulphur   |  |  | 2 each                       |  |
| 8                          | Describe a method which can be used to separate two components with different solubilities in a solvent X.  |  |  | 2                            |  |
| 9                          | How can you separate a mixture of acetone and methanol? Give the principle behind it.   |  |  | 2                            |  |
| 10                         | Which of the following has higher boiling point and why?<br>2-methylpentane, n-hexane, 2,3-dimethyl butane  |  |  | 2                            |  |
| 11                         | How will you convert ethyne and hexane to benzene?  |  |  | 2                            |  |
| 12                         | Complete the reaction<br>a) $\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)_2 + \text{O}_3 \longrightarrow$<br>b) $\text{C}_4\text{H}_{10} \xrightarrow{773\text{K}}$   |  |  | 1 each                       |  |

|    |   |           |
|----|---|-----------|
|    | c) $\text{CH}_3\text{C}\equiv\text{CCH}_3 + \text{H}_2 \xrightarrow{\text{Na/liq.NH}_3}$  |           |
| 13 | Convert<br>a) isopropyl bromide to n-propyl bromide<br>b) acetic acid to methane<br>c) methane to ethene<br>d) ethanoic acid to benzene<br>e) ethane to ethyne<br>f) 1-bromopropane to 2-bromopropane<br>g) propyne to propanone  | 1<br>each |
| 14 | How will you distinguish between ethane and ethylene?   | 2         |
| 15 | How does ethylene react with<br>a) bromine<br>b) alkaline $\text{KMnO}_4$<br>c) hydrogen in presence of Lindlar's catalyst  | 1<br>each |
| 16 | Which alkene on ozonolysis produces propanone only? Write the reaction involved.  | 2         |
| 17 | An alkene A on ozonolysis gives a mixture of ethanol and pentan-3-one. Identify A. Write the structure and IUPAC name of A. Also write the equations involved.  | 2         |
| 18 | Give the chemistry behind detection of nitrogen and halogen qualitatively.  | 3         |
| 19 | Find whether the following compounds are aromatic or not?<br><br>  | 1<br>each |
| 20 | Addition of HBr to propene yields 2-bromopropane, while in presence of benzoyl peroxide, the same reaction yields 1-bromopropane. Explain and give the mechanism.   | 3         |
| 21 | What happens when?<br>a) Benzene is treated with methyl chloride in presence of an. $\text{AlCl}_3$<br>b) Benzene is treated with a mixture of con. $\text{HNO}_3$ and con. $\text{H}_2\text{SO}_4$<br>c) Pentane is heated with HCl in presence of an. $\text{AlCl}_3$ .   | 3         |
| 22 | An alkyl halide $\text{C}_5\text{H}_{11}\text{Br}$ (A) reacts with ethanolic solution of KOH to give an alkene (B) having molecular formula $\text{C}_5\text{H}_{10}$ . B reacts with $\text{Br}_2$ in $\text{CCl}_4$ to give a compound C. C on dehydrobromination with sodamide gives a compound D having molecular formula $\text{C}_5\text{H}_8$ . D on complete hydrogenation yields a straight chain alkane. Identify A, B, C and D. Give the equations involved. | 5         |